6. Chemical Properties

6 • 1 Rate of Water and Moisture Absorption

The rate of water absorption is

ASTM – D570, after 24 hours in water $0.23\sim0.26\%$

The rate of water and moisture absorption of Iupilon / NOVAREX is as shown in Fig. $6 \cdot 1 - 1$.

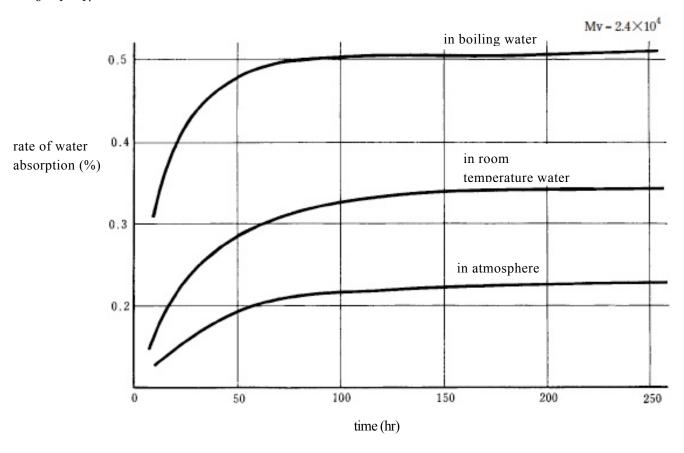


Fig. 6 · 1—1 Water absorption curve of Iupilon / NOVAREX (ASTM-D570)

Also, the rate of water absorption measured according to JISK 6771-1958 was

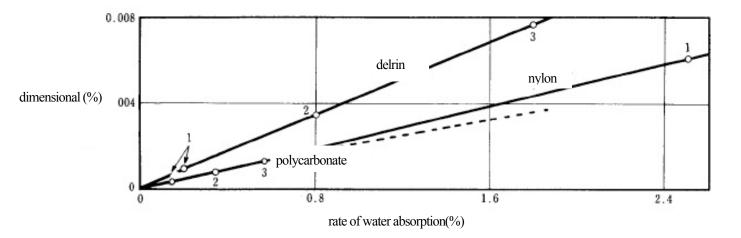
After 24 hours in 50°C water 0.29% After 24 hours in 75°C water 0.39%

This method of measuring the rate of water absorption is not necessarily based on rate of water absorption of zero.

In view of this, if the moisture of these moisture-absorbed samples is determined by the Karl Fischer method, the coefficient of water absorption becomes about $0.05 \sim 0.10\%$ larger than the above values.

The dimensional change of polycarbonate by water and moisture absorption is as shown in Fig. $6 \cdot 1-2$.

A comparison with other resins is shown in Table 6 • 1-1, from which it can be seen that the rate of water absorption of Iupilon / NOVAREX is very small and consequently, one of its features is that dimensional change due to this is very small.



1: room temperature, 50% RH, 2: in room temperature water, 3: in boiling water Fig. $6 \cdot 1-2$ Dimensional change by water absorption

Table 6 • 1-1 Rate of water absorption of various resins ASTM - D570 (after 24 hours)

Name of substances	Rate of water absorption (%)			
Low density polyethylene	< 0.01			
Polycarbonate	0.23			
Rubber modified PVC	0.07~0.2			
Cellulose propionate	1.2~2.8			
High impact strength ABS	0.3			
High impact strength ethylcellulose	0.8~2.0			
Ethylcellulose	1.2~2.0			
Soft cellulose acetate	2.3~6.5			
Soft cellulose butyrate	0.9~1.3			
Rigid cellulose butyrate	1.3~1.8			
Cellulose acetate	1.9~4.0			
Teflon	0.00			
Glass fiber reinforced nylon	0.7~1.4			
Polyethylene	0.04~0.08			
A B S	0.2			
Rigid cellulose acetate	1.6~3.8			
Nylon	1.2~2.3			
Acrylate	0.2~0.4			
Polyacetal	0.41			
Polypropylene	< 0.01			

6 · 2 Gas Permeability

Water vapor permeability of Iupilon film is as shown in Table 6 • 2-1.

Table 6 • 2-1

Film thickness		Rate of water vapor					
(mm)	Testing method permeability						
(111111)		(g∕m²•day)					
0.20	JIS-Z 0208, tempt. 40°C, humidity90%	23.3					
0.08	ASTM-E 96-53T, tempt. 50°C, humidity50%	46-57					
0.05	"ASTM-E 96-53T, tempt. 50°C, humidity 50%	69					
0.02	JIS-Z 0208, tempt. 40°C, humidity90%	195					

Gas permeability of Iupilon is as shown in Table $6 \cdot 2 - 2$ and Fig. $6 \cdot 2 - 1$. The relation between gas permeability constant and 1/T is shown by the straight line but it has a refractive point between $0 \sim 20$ °C in case of gas with relatively big constant.

Also, the relation between gas permeability of Iupilon and sample thickness is as shown in Table $6 \cdot 2 - 3$. It depends on the kind of gas but shows almost a constant value until the thickness of 300 µm except for CO₂.

Table $6 \cdot 2 - 2$ Gas permeability of Iupilon film (ASTM-D1434)

Townsustans (°C)		Gas per	rmeability co	nstan t *	
Temperature (°C)	H_2	Не	N_2	02	CO ₂
90			20. 6		
80			16. 2		
70			12. 3	46. 4	
60	533.	315.	9. 7	39. 2	210.
50	352.	237.	6. 1	31. 9	160.
40	208.	197.	4. 9	23. 8	127.
30	165.	144.	3. 2	18. 1	103.
20	120.	110.	2. 4	14. 3	72.
0	62.	77.	1. 4	8. 4	48.
—10	47.	48.	0. 8	5. 8	31.
—20			0. 4	4. 0	
—25				3. 3	
—30	34.	34.		2. 9	19.
—35	24.	25.		2. 3	11.
-40	17.	18.		2. 1	5. 0
—4 5	12.	13.		1. 3	2. 1
—50	7.	8.		1. 0	1. 3
— 55				0. 9	
-60				0. 8	

* Unit cm³ · cm/cm² · s · cmHg · 10¹¹, pressure difference 1 atm, thickness 50 μ

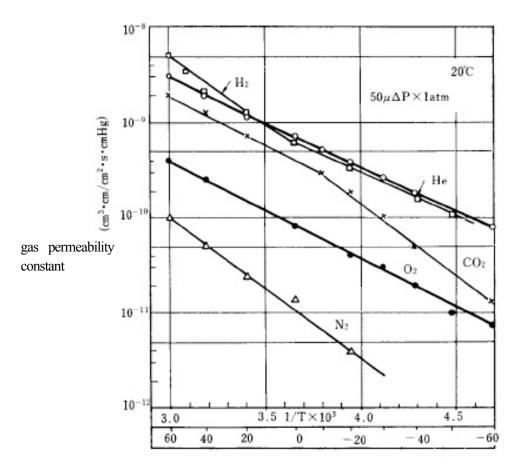


Fig. $6 \cdot 2 - 1$ Gas permeability of Iupilon

The relation between gas permeability of Iupilon and thickness is as shown in Table 6 • 2-3.

Table 6 • 2-3 Relation between gas permeability of Iupilon and thickness

Thickness (μm)	Gas permeability constant⊁						
IIII CKIIESS (μ III)	H ₂ N ₂		02	CO_2			
50	120.	2. 4	14. 3	72			
100	109.	2. 5	14. 2	56			
200	105.	2. 3	12. 2	26			
300	105.	2. 4	12. 8	24			
600	114.		9. 1				

* Unit cm 3 · cm/cm 2 · s · cmHg · 10^{11} , pressure difference 1 atm, tempt. 20° C

The values of water vapor and gas permeability for other resins are shown in Table $6 \cdot 2 - 4$.

	Rate of water		Gas permeabilit	ty constant * *			
	vapor permeability *	H_2	02	N_2	CO_2		
Polyethylene	4. 3	10. 07	3. 47	1. 22	12. 5		
Mylar	9. 5	0. 80	0. 04	0. 008	0. 20		
Polyvinyl chloride	22. 1	3. 8	0 70	0 00	1. 78		
Nylon	43. 8	1. 0	0. 78 5. 8	0. 28 2. 79	0. 31		
Polystyrene	45	33. 1	ე. გ	Z. 19	10. 5		

Table $6 \cdot 2 - 4$

ASTM • E96 – 53T, temperature 37.8°C, water in cup, outside humidity 0%, unit g/m2 • day/0.1mm

6 • 3 Chemical Resistance

Chemical resistance of Iupilon / NOVAREX at room temperature and 75°C is shown in Table $6 \cdot 3-2$.

As Iupilon / NOVAREX is a polyester, there is a limit to resistance against alkaline chemicals as they cause hydrolysis of the ester bond. As a general rule, it has a strong tendency of swelling and dissolving in aromatic hydrocarbons and chlorinated hydrocarbons.

Also, the ester bond is broken at high temperature by alcohols and phenols.

At a high temperature $(60\sim100^{\circ}\text{C})$, it is degraded when left standing for a long time even in pure water, as mentioned in paragraph $4 \cdot 6$ and in view of this, the effect on chemical resistance at 75°C can not be overlooked.

As a general rule, the phenomenon of stress cracking or solvent corrosion is present in plastics, by which cracks are formed when it comes in contact with some solvent or its vapor under stress.

Precaution must be taken as this phenomenon is also present in Iupilon / NOVAREX. Solvents to which Iupilon / NOVAREX is particularly sensitive when under stress are carbon tetrachloride, hexane, acetone and alcohols, and some examples are given in Fig. $6 \cdot 3 - 1$.

Also, the limiting tensile strain at which stress cracking is produced by carbon tetrachloride, acetone and hexane is 0.5, 0.35 and 0.8%, respectively, while compression strain does not cause such stress cracking.

In addition, it is known that wide-range substances, for example plasticizers, grease, rubber, and various oils etc. also cause the solvent corrosion, and some examples are given in Table $6 \cdot 3 - 1$.

It is necessary to test the solvent corrosivity of the used substance at the time of design of practical articles.

^{* *} Temperature 30° C, unit cm³ · cm/cm² · s · cmHg · 10^{10}

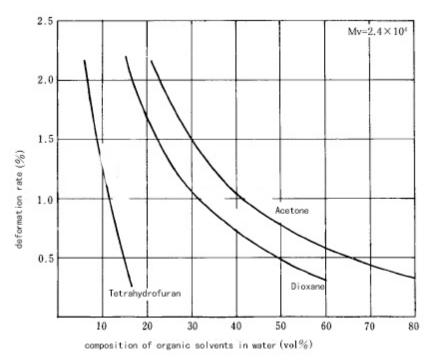


Fig. 6 • 3 – 1 Solvent cracking curve under stress of Iupilon / NOVAREX (Tempt. 25°C, test piece 152×12 . 7×6 . 4mm)

A combination of flexural strain that does not produce crack and solvent composition was plotted for the case the product was immersed in mixed solvents of various compositions for 1 minute in a condition various flexural strains were given.

Table $6 \cdot 3 - 1$ Solvent corrosion of Iupilon [Unit MPa (kgf/cm²)

	Solvent		Substances		Solvent		
Substances	corrosion				corro		
	R. T	75°C		R.	T	75°C	
Elevible DVC (with 6004 placticizer)		3. 9	Cutting oil (Vamirages)	29.	4	3. 9	
Flexible PVC (with 60% plasticizer)	7. 8	(40)	Cutting oil (Komirogen)	(30	0)	(40)	
Grease (Albania No. 2)	(80)	9. 8	Cutting oil (Komirogen 30% aqueous	29.	4	7. 8	
Grease (Albania No. 2)	33. 3	(100)	solution)	(30	0)	(80)	
Crosse (Albania No. 2 love toman 1982)	(340)		Cutting ail (Dramag ail D)	29.	4	3. 9	
Grease (Albania No. 2 low temp. use)		15. 7	Cutting oil (Dromas oil B)		0)	(40)	
Grease (beacon No. 325)	8. 8	(160)	Cutting oil (Dromas oil 60% aqueous	29.	4	7. 8	
Glease (beacon no. 323)	(90)	15. 7	solution)	(30	0)	(80)	
Grease (silinon SH7071)	33. 3	(160)	Salad oil	16.	7	7. 8	
Grease (sililoii SH/0/1)	(340)		Salad oil		0)	(80)	
Not wal mikhan	25. 5	7. 8	Antifuncal oil (DA - 50)	29.	4	16. 7	
Natural rubber	(260)	(80)	Antifungal oil (PA — 50)	(30	0)	(170)	
Nacarana milihar (A)	24. 5	7. 8	Antifuncal ail (Taiona)	29.	4	19. 6	
Neoprene rubber (A)	(250)	(80)	Antifungal oil (Taiana)	(30	0)	(200)	
Nagarana milihan (D)	8. 8	3. 9	Autifum and aid (CDC 20)	19.	6	8. 8	
Neoprene rubber (B)	(90)	(40)	Antifungal oil (CRC—36)		0)	(90)	
D + l = l·l = :	21. 6	7. 8	G1(A)	21.	6	3. 9	
Butyl rubber	(220)	(80)	Sealant (A)	(22	0)	(40)	

Ethylene propylene rubber	12. 7 (130)	3. 9 (40)	Sealant (B)	29. 4	7. 8 (80)
Silicon rubber (A)	32. 4 (330)	15 7	Epoxy cast agent	(300) 20. 6	3. 9 (40)
Silicon rubber (B)	25. 5 (260)	15 7	Silicon cast agent	(210)	11. 8 (120)
Rock materials	3. 9 (40)	3. 9	Blank	39. 2 (400)	22. 6 (230)

Testing method: Let the substance touch the part where maximum flexural stress is generated by constant strain method, and leave it at normal temperature for 48 hours, at 75°C for 3 hours, then calculate the strain quantity that crack is not generated and convert it to stress.

Table 6 · 3-2 Chemical resistance of Iupilon / NOVAREX at room temperature and 75°C Size of test piece: 63.5mm length, 12.7mm width, 3.2mm thickness Strength before tes: Tensile strength 58.1MPa (592khf/cm2), Izod impact strength (with notch) 910J/m

Figures in () indicate values at 75°C and all are values after immersion for 30 days.

Name of chemicals		Change in weight	Change in tensile strength %	Impact value J/m	Change in appearance
Inorganic salts					
Sodium chloride	15%	+0.14	-4. 4	890	No change
	(15%)	(+0. 22)	(+1. 52)	(-)	Small number of cracks
Sodium sulfide	15%	+0.13	-5. 4	930	No change
	(15%)	(+0.15)	(+3.55)	(-)	Haze on surface, small number of cracks
Potassium chloride	15%	+0. 14	$-3. \ 2$	950	No change
Potassium nitrate	15%	+0. 15	-3. 6	920	No change
1	(15%)	(+0. 21)	(+1. 18)	(-)	Small number of cracks
Potassium dichrom	ate 15%	+0. 15	-2. 9	950	No change
	(15%)	(+0. 31)	(-0. 7)	(-)	Small number of cracks
Calcium chloride	15%	+0.14	-4. 4	930	No change
	(15%)	(+0. 15)	(+0.7)	(—)	Small number of cracks
Sodium sulfate	10%	(+0. 25)	(+1. 5)	(—)	Small number of cracks
Inorganic acids					
Hydrochloric acid	35%	+0. 13	-8. 5	56	Cracks
	(15%)	(+0. 18)	$(+2. \ 4)$	(-)	Cracks
	10%	+0.10	-2. 7	920	No change
	(5%)	(+0. 38)	(十1. 5)	(-)	Cracks
Sulfuric acid	98%	_	_	_	Whiten, dissolve
	90%	+5.16	-1. 9	890	Milk white
_	85%	-0. 17	$-5 \cdot 4$	910	No change
	80%	-0.30	-4. 1	920	No change
	(80%)	(-0.78)	(+1. 5)	(-)	No change
	50%	-0. 13	−3. 0	860	No change
	(50%)	(-0. 10)	(+3. 6)	(—)	No change
	10%	-0. 13	-6. 0	880	No change'
	(10%)	(+0. 19)	(+0. 8)	(-)	No change
Nitric acid	60%	十1. 41	−9. 2	920	Discolor to yellow
	(60%)	(-)	(-)	(-)	Shape destroyed

30%	+0. 25	-6. 6	640	Discolor to yellow
(30%)	(+1. 06)	(-7. 9)	(—)	Opaque, yellow
10%	+0.14	$-2. \ 0$	910	No change
(10%)	(+0, 33)	(0)	(-)	Transparent yellow
Orthophosphoric acid 100%	-0.25	+0. 8	960	No change
(100%)	(-0. 12)	$(+3. \ 2)$	(-)	Transparent light yellow
10%	+0.09	-3.4	900	No change
(10%)	(+0. 24)	(+2. 7)	(-)	Cracks

		Chang	_	Change in tensile	Impact value	
Name of chemic	cals	weig	_	strength	J/m	Change in appearance
		%		%		
Chromic acid	10%	+0.	25	— 3. 7	950	No chang
Hydrogen fluoride	conc.	+1.	42	-9 . 3	1000	No charge
Sulfuric acid 5	0 vol%	+ 5.	02	— 19. 0	88	Dissolve
Nitric acid 50	vol%	10.	02	10. 0	00	D1000140
Sulfuric acid	Ü	+0.	11	-4 . 4	920	No change
Chromic anhydric	de 150g	10.	17	7. 7	320	ino onange
Water 1.51						
Inorganic alkali						
Sodium carbonate	15%	+0.	13	- 3. 2	870	No change
	(15%)	(-0.	79)	(+2. 0)	(—)	Cracks
Sodium hydroxide	(10%)	(-2.	92)	(-1. 4)	(—)	Surface haze, cracks
	5%	+0.	03	-4 . 4	950	No change
	(5%)	(-0.	17)	(-0. 8)	(—)	Surface haze, cracks
	1%	+0.	05	-4 . 6	990	No change
Milk of lime	10%	+0.	04	- 5. 1	970	No change
	(5%)	(-0.	46)	(+2. 7)	(—)	Cracks
Organic acids	•					
Acetic acid	100%	+3.	25	— 7. 3	660	Turbid
	70%	+0.	15	— 2. 0	990	No change
	(70%)	(+0.	86)	(-40. 5)	(—)	Turbid
	50%	+0.	14	-4 . 7	970	No change
	(50%)	(+0.	64)	(-33. 5)	(—)	Turbid
	10%	+0.	21	-4 . 1	950	No change
	(10%)	(+0.	48)	(-44. 1)	(—)	Cracks
Formic acid	97%	+1.	92	—12 . 7	970	No change
	70%	+0.	68	-6 . 4	920	No change
	(70%)	(+1.	91)	(-77. 5)	(—)	Cracks
	(40%)	(+0.	86)	(-75. 7)	(—)	Cracks
	10%	+0.	28	-4 . 7	950	No change
Maleic acid	10%	+0.	07	-4 . 1	920	No change
Benzoic acid	10%	+0.	21	-6 . 6	990	No change
Lactic acid	10%	+0.	09	— 1. 4	930	No change
	(10%)	(+0.		(+4. 9)	(—)	Cracks
	1%	+0.		-4 . 4	920	No change
Oxalic acid	(30%)	(+0.		(+2. 7)	(—)	Cracks
	10%	+0.	12	-4 . 1	940	No change

Picric acid	2%	+0. 13	- 6. 7	960	No change
Petroleum components					
n—Hexane		+0. 07	- 5. 4	930	No change
Solvent naphtha					Turbid, swells
Cyclohexane		+0. 07	-4 . 4	920	No change
		(—)	(-)	(-)	Dissolve completely

	Change in	Change in tensile	Impact value	
ıls	weight	strength	•	Change in appearance
	%	%	J/ 111	
	+0. 003	-2. 4	950	No change
	+0. 08	-4 . 4	930	No change
	(-0. 07)	(+0. 2)	(—)	Cracks
	+0. 15	— 7. 1	970	No change
°C)	(+0. 49)	(-0. 3)	(—)	No change
	_	_	_	Swell, dissolve
	_	_	_	Swell, dissolve
	+0. 003	— 5. 2	930	No change
	+0. 003	— 5. 9	940	No change
	(-0. 05)	(+2. 0)	(—)	No change
	+0. 003	-4 . 7	950	No change
	+0. 03	- 0. 9	940	No change
	(+0. 02)	(+0. 7)	(—)	No change
	+0. 018	+0. 8	970	No change
	+0. 017	-4 . 1	940	No change
	+0. 007	— 2. 9	960	No change
	(+0. 01)	(+0. 7)	(—)	No change
	+0. 07	— 1. 7	940	No change
omatics)	+0. 009	— 1. 9	920	No change
	(-0. 02)	(+1. 9)	(-)	No change
•				
	+0. 08	- 5. 2	930	No change
	+0. 07	— 5. 9	930	No change
	(-0, 13)	(+2, 5)	(—)	No change
				No change
				in oursings
	+1. 47	-21. 5	1000	Cracks on surface
100%	+0. 50	-11. 2	960	No change
				Turbid
				Cracks
, ,				No change
	+0. 12			No change
				Whitens to translucet
				Whitens to translucent
				No change
				No change
	(+0. 04)	(+1. 69)	(-)	No change
	omatics) 100% (100%) (50%)	9% +0. 003 +0. 08 (-0. 07) +0. 15 (+0. 49) +0. 003 +0. 003 (-0. 05) +0. 003 (+0. 02) +0. 018 +0. 017 +0. 007 (+0. 01) +0. 07 (+0. 01) +0. 07 (+0. 01) +0. 07 (-0. 13) +0. 08 +1. 47 +0. 08 100% (100%) (+3. 41) (+1. 18) (+1. 39)	96 96 +0. 003 -2. 4 +0. 08 -4. 4 (-0. 07) (+0. 2) +0. 15 -7. 1 -7	Second Strength Strength

Glycerine	— 0. 07	— 6. 9	960	No change
	(+0. 06)	(+0. 5)	(-)	No change
Ethyl ether	+15. 4	— 55. 7	1000	Milky white
Acetone	_	_	_	Turbid, crumbles
Name of chemicals	Change in weight %	Change in tensile strength %	Impact value J/m	Change in appearance
Methyl ethylketone				Turbid, crumbles
Ethyl acetate				Turbid, swells
Carbon tetrachloride	+7. 9	-9 . 6	960	Turbid
Carbon disulfide	+2. 0	- 5. 4	930	Turbid
Triethanol amine	— 0. 21	— 3. 6	930 910	Cracks on surface
Terpene oil	— 0. 02	— 3. 7	910	No change
	(+1. 69)	(+2. 0)	(—)	Translucent turbid
Camphor oil	— 0. 01	-4 . 3	960	No change
	(+1. 53)	(+0. 2)	(—)	Translucent turbid
Others	1			1
Toilet soap 3%	+0. 07	- 5. 2	960	No change
N. I. I. Ook	+0. 07	—4 . 9	960	No change
Neutral soap 3%	(+0. 25)	(+2. 7)	(—)	Cracks
Bleaching powder 2. 5%	+0. 05	- 5. 4	970	No change
Hydrogen peroxide 70%	+0. 34	— 3. 9	820	Light yellow, tanslucent
309	6 +0. 15	— 0. 7	880	Light yellow
Formal in 2006	+0. 17	-4 . 2	910	No change
Formalin 30%	(+0. 83)	(+4. 9)	(—)	Cracks
Black tea	(+0. 27)	(+2. 0)	(-)	Cracks
Chlorine	+1. 5	- 0. 2	910	Discolor to yellow